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BG—24—2016

FACULTY OF SCIENCE

B.Sc. (BT) (First Semester) EXAMINATION

OCTOBER/NOVEMBER, 2016

(Revised Course)

BIOTECHNOLOGY

Paper BTT-1.4

(Fundamentals of Chemistry)

(Friday, 25-11-2016)

Time : 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—80

N.B. :— (i) All questions are compulsory.

(ii) Draw neat well labelled diagrams wherever necessary.

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|----|-----|---|----|
| 1. | (a) | Describe in detail Planck's quantum theory. | 10 |
| | (b) | Explain electronic configuration of atoms. | 10 |
| 2. | (a) | Describe in detail postulates of Bohr's atomic model. | 8 |
| | (b) | Describe in detail VSEPR theory. | 7 |
| | | <i>Or</i> | |
| | (a) | Explain lattice energy. | 8 |
| | (b) | Describe van der Waals interactions. | 7 |
| 3. | (a) | Explain colligative properties of solutions. | 8 |
| | (b) | Describe in detail dipole movement. | 7 |
| | | <i>Or</i> | |
| | (a) | Discuss pH, buffer and their applications. | 8 |
| | (b) | Describe Ionic equilibrium in solution. | 7 |
| 4. | (a) | Explain zero, first and second order kinetics. | 8 |
| | (b) | Discuss catalysis and elementary enzyme reactions. | 7 |

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Or

- (a) Explain first law of thermodynamics. 8
- (b) Describe in detail Hess law. 7
5. Write short notes on (any *three*) : 15
- (a) Electronegativity
- (b) Hybridization
- (c) Common ion effect
- (d) Tautomerism
- (e) Heat of formation.

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